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## 次世代型 PET 化学リサイクルに向けた CO<sub>2</sub> 統合プロセスの設計

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### Design of CO<sub>2</sub>-integrated processes for next-generation chemical recycling of PET

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### 研究概要

Plastics such as polyethylene terephthalate (PET) offer excellent performance to improve efficiency in daily life, but their accumulation has become a serious environmental concern. Developing effective strategies to address this “white pollution” is therefore an urgent challenge. Conventional waste-management approaches rely heavily on thermal and electrical energy, producing CO<sub>2</sub> as a major “black pollution” byproduct. Given the rapid rise in atmospheric CO<sub>2</sub> levels, expanding its utilization is essential for achieving net-zero emissions. A process that valorizes both PET waste and CO<sub>2</sub> simultaneously presents a promising solution aligned with carbon-neutrality and plastic-recycling policies.

Our research focuses on developing reaction systems that utilize CO<sub>2</sub> for the efficient degradation of PET. Because PET depolymerization is reversible, shifting the reaction equilibrium toward the products is crucial. Transesterification can proceed effectively in the presence of a trapping agent that captures the byproduct, thereby driving the reaction forward. CO<sub>2</sub> plays a key role by enabling the in-situ formation of this trapping agent.

This study aims to design an effective catalyst capable of activating CO<sub>2</sub> to cleave the C–O bond in the ester moiety of PET. Using CO<sub>2</sub> in catalytic PET degradation provides a dual benefit: mitigating plastic pollution while converting CO<sub>2</sub> into valuable materials, offering a pathway to next-generation chemical recycling. With an estimated recycling efficiency of at least 70%, depolymerizing 1 kg of PET could potentially consume approximately 160 g of CO<sub>2</sub> (equivalent to ~81.3 L), demonstrating significant potential for combined waste valorization.